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## Reaction of 2-(2-Oxo-1,2-dihydro-3*H*-indol-3-ylidene)acetic Acids Esters with Phenylhydrazine

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**Abstract**—2-(2-Oxo-1,2-dihydro-3*H*-indol-3-ylidene)acetic acids esters reacted with phenylhydrazine yielding products of the regioselective addition of the latter in the  $\alpha$ -(C<sup>2</sup>)-position of the exo ethylene bond, (2-oxo-2,3-dihydro-1*H*-indol-3-yl)(2-phenylhydrazino)acetic acids esters.

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A reaction of 3-(2-oxo-2-X-ethylidene)-1*H*-indol-2ones (**I**, X = Alk, Ar, Ht) with hydrazines under mild conditions is known to provide 2',4'-dihydrospiro-[indol-3,3'-pyrazol]-2(1*H*)-ones **II** [1–7] (Scheme 1). Compounds **II** form apparently as a result of an intramolecular spiroheterocyclization of intermediate 3-(2-hydrazono-2-X-ethylidene)-1*H*-indol-2-ones **III** at the 3-exo ethylene bond into the  $\beta$ -position (C<sup>3</sup>) of the indole ring. In some cases stable hydrazones (**III**, X = Ar) were isolated and identified [5, 8] (Scheme 1).

At treating 3-(2-aryl-2-oxoethylidene)-1H-indol-2ones (I, X = Ar) with 2-hydrazino-1*H*-benzimidazole formed both the corresponding hydrazones (III,  $R^1$  = Me, Ac) and products of cyclization at the carbonyl group of lactam, 3-aryl-1-(1H-benzimidazol-2-yl)-1H-pyridazino[3,4-b]indoles (**IV**, R<sup>1</sup> = H) [8] (Scheme 1). It was reported that the heating of phenacylideneoxindole (I,  $R^1 = R^2 = H, X = Ph$ ) with hydrazine resulted in reduction to oxindole and elimination of phenylglyoxal hydrazone [1, 2]. The hydrazine attack occurred here evidently at the electron-deficient site  $\alpha$ -(C<sup>1</sup>) of indolinone I. Note that some spiroheterocycles II exhibited antimicrobial activity [7], and compounds III and IV, insecticidal action [8]; these facts demonstrated the practical importance of products of N-nucleophilic transformations of ylideneoxindoles I.

We recently showed that unlike X-acyl derivatives of ylideneoxindoles (I, X = Alk, Ar, Ht) similar in structure

esters of 2-(2-oxo-1,2-dihydro-3H-indol-3-ylidene)acetic acids ( $\mathbf{I}, \mathbf{X} = \mathbf{OAlk}$ ) differently reacted with hydrazine. Thus in acetic acid formed a mixture of products of  $\alpha$ -addition to the exo ethylene bond with respect to the ester group (followed by recyclization), 3,3a,5,9b-tetrahydro-1*H*-pyrazolo[3,4-c]quinoline-1,4(2*H*)-dione (V) and of  $\beta$ -addition, 1-acetyl-5'H-spiro[indole-3,3'-pyrazolidine]-2,5'(1H)-dione (VI) [9, 10]. The addition of aromatic amines to the exocyclic multiple bond of substrates (I, X = OAlk) in contrast to hydrazine occurred regioselectively into the  $\alpha$ -position leading to the formation of 2-arylamino-2-(2-oxo-2,3-dihydro-1H-indol-3yl)acetic acid esters VII [11, 12] (Scheme 1). According to our findings the attack of mono- and binucleophiles is directed prevailingly on the electrophilic site  $\alpha$ -(C<sup>2</sup>) of oxindolylideneacetates (I, X = OAlk) [9, 12–14].

Thus the published data show that to the attack of hydrazines (and amines) are subjected four electrophilic sites of ylideneoxindoles (I, X = Alk, Ar, Ht, OAlk): at the atioms C<sup>2</sup> (NC<sup>2</sup>=O), C<sup>3</sup> ( $\beta$ -position of 3-exo ethylene bond), C<sup>1'</sup> ( $\alpha$ -position), and C<sup>2'</sup> (XC<sup>2'</sup>=O); it is difficult a priori predict the direction of the nucleophilic attack.

We established that by treating with phenylhydrazine 2-(2-oxo-1,2-dihydro-3*H*-indol-3-ylidene)-acetic acids esters **Ia–If** at boiling the mixture in ethanol esters of (2-oxo-2,3-dihydro-1*H*-indol-3-yl)-(2-phenylhydrazino)-acetic acids **VIIIa–VIIIf** were obtained in a preparative yield (Scheme 2). Compounds **VIII** result from

a regioselective addition of the primary amino group of phenylhydrazine to the *exo* ethylene bond of substrate **I** not in the  $\beta$ -(C<sup>3'</sup>)-position as might be expected by analogy with the similar reactions of acylmethyleneoxindoles (**I**, X = Alk, Ar, Ht) but in the  $\alpha$ -(C<sup>2</sup>)- position with respect to the ester moiety. Compounds obtained **VIIIa–VIIIf** are colorless or off-yellow crystalline substances insoluble in water, sparingly soluble in the common organic solvents, and readily soluble in DMF and DMSO.

Scheme 1.



 $R^1 = H$ , Alk, All, Ac;  $R^2 = H$ , Alk, All, AlkO, Ac, Hlg; X = Alk, Ar, Ht;  $R^3 = H$ , Ph, Ht, Ac.



Scheme 2.



I, VIII, Alk = Me:  $R^1$  = H,  $R^2$  = H (a), Br (b);  $R^1$  = Ac,  $R^2$  = H (c); Alk = Et:  $R^1$  = H,  $R^2$  = H (d), Br (e);  $R^1$  = Ac,  $R^2$  = H (f). RUSSIAN JOURNAL OF ORGANIC CHEMISTRY Vol. 43 No. 10 2007

In the IR spectra of compounds **VIII** are present well consistent with the structure wide bands of the stretching vibrations of amino groups from the phenylhydrazine fragment and of the amide group of lactam at 3130–3293 cm<sup>-1</sup>, and also absorption bands of carbonyl groups of ester (1720–1732 cm<sup>-1</sup>) and lactam (1673–1696 cm<sup>-1</sup>).

The proton signals in the <sup>1</sup>H NMR spectra of indolinones VIIIa-VIIIf appear in a double set, in particular, pairs of signals are present from two vicinal methine protons of the fragment C<sup>2</sup>H–C<sup>3</sup>H at  $\delta$  4.11–4.38 and 4.35–4.73 ppm. The character of these signals indicates the presence of at least two diastereomers, and the findings obtained are in agreement with the spectral characteristics of 2-arylamino-2-(2-oxo-2,3-dihydro-1Hindol-3-yl)acetic acids esters VII [12]. The presence of a pair of coupled methine protons permits a rejection of the regioisomeric structure of 2-oxo-3-(2-phenylhydrazino)-2,3-dihydro-1*H*-indol-3-ylacetic acids esters **IX** containing a methylene group  $CH_2$  (Scheme 2). It was not possible to exclude preliminary the formation of the latter taking into account published data, in particular, on the  $\beta$ -addition of amines to the exo ethylene bond of vlideneoxindoles I [15].

Thus the reaction of compounds **Ia–If** with hydrazine proceeds as a regioselective addition of the latter to the exo ethylene bond into the  $\alpha$ -position with respect to the ester group. The changed nucleophiles addition direction, in particular, that of phenylhydrazine, into the  $\alpha$ -C<sup>2</sup>-position of oxindolylideneacetates (**I**, X = OAlk) [11–14] unlike the usual  $\beta$ -C<sup>3'</sup>-attack of 3-acylmethylene-1*H*-indol-2-ones (**I**, X = Alk, Ar, Ht), probably is caused by stronger electron-acceptor effect of the ester group in the former substrates compared to the het(aroyl) moiety of the latter.

## **EXPERIMENTAL**

IR spectra of compounds **VIII** were recorded on a spectrophotometer Specord M-80 from mulls in mineral oil. <sup>1</sup>H NMR spectra of indolinones **VIII** were registered on a spectrometer Bruker DRX-500 (500.13 MHz) in DMSO- $d_6$ , internal reference TMS. Mass spectrum of compound **VIIIa** was measured on a Finnigan MAT INCOS-50 instrumen at the direct admission mode (electron impact). The reactions progress was monitored and the homogeneity of compounds **VIIIa–VIIIf** was checke by TLC on Silufol UV-254 plates in a system benzene–ether–acetone, 10:9:1, development in iodine vapor. Initial esters **Ia–If** were obtained by procedures [16–18]. (2-Oxo-2,3-dihydro-1*H*-indol-3-yl)-(2-phenylhydrazino)acetic acids esters VIIIa–VIIIf. To a solution of 10 mmol of esters Ia–If in 70–100 ml of ethanol was added at stirring 1.08 g (10 mmol) of phenylhydrazine, and the mixture was boiled for 0.5–2 h (TLC monitoring). The separated precipitate was filtered off and recrystallized from ethanol.

Methyl (2-oxo-2,3-dihydro-1H-indol-3-yl)(2-phenylhydrazino)acetate (VIIIa). Yield 2.46 g (79%), mp 148-149°C. IR spectrum, v, cm<sup>-1</sup>: 3287 (C<sub>6</sub>H<sub>5</sub><u>NH</u>NH), 3130– 3185 (<u>N<sup>1</sup>H</u>CO, C<sub>6</sub>H<sub>5</sub>NH<u>NH</u>), 1728 (C=O<sub>ester</sub>), 1696 (N<sup>*I*</sup>H<u>CO</u>), 1618, 1597, 1466. <sup>1</sup>H NMR spectrum, δ, ppm: 3.65 s (3H, COOCH<sub>3</sub>), 3.82 s (3H, COOCH<sub>3</sub>), 4.18 d, 4.32 d (2H, C<sup>3</sup>H, C<sup>2</sup>H, J 4.3 Hz), 4.47 d, 4.65 d (2H, C<sup>3</sup>'H, C<sup>2</sup>H, J 5.6 Hz), 5.22 br.s (2H, 2C<sub>6</sub>H<sub>5</sub><u>NH</u>NH), 6.62 s (1H,  $C_6H_5$ NHNH), 6.77–7.36 group of signals (18H, 2C<sub>6</sub>H<sub>4</sub>, 2C<sub>6</sub>H<sub>5</sub>), 7.38 s (1H, C<sub>6</sub>H<sub>5</sub>NH<u>NH</u>), 10.26 s (1H, N<sup>1</sup>'H), 10.64 s (1H, N<sup>1</sup>'H). Mass spectrum, m/z ( $I_{rel}$ , %): 311 (14)  $[M]^+$ , 252 (3)  $[M - \text{COOCH}_3]^+$  or  $[C_{15}H_{14}N_{3}O]^{+}$ , 203 (37)  $[M - C_{6}H_{5}NHNH_{2}]^{+}$  or  $[C_{11}H_{9}NO_{3}]^{+}$ , 172 (35)  $[M - C_{6}H_{5}NHNH_{2} - OCH_{3}]^{+}$  or  $[C_9H_6NO_2]^+$ , 147 (40)  $[C_9H_9NO]^+$  or  $[C_8H_5NO_2]^+$ , 144  $(34) [M - C_6H_5NHNH_2 - COOCH_3]^+ \text{ or } [C_9H_6NO]^+, 133$ (100) [oxindole =  $C_8H_7NO$ ]<sup>+</sup>, 108 (38) [ $C_6H_5NHNH_2$ ]<sup>+</sup>, 92 (72) [C<sub>6</sub>H<sub>5</sub>NH]<sup>+</sup>, 77 (61) [C<sub>6</sub>H<sub>5</sub>]<sup>+</sup>. Found, %: C 65.29; H 5.72; N 13.34. C<sub>17</sub>H<sub>17</sub>N<sub>3</sub>O<sub>3</sub>. Calculated, %: C 65.58; H 5.50; N 13.50. M 311.33.

Methyl (5-bromo-2-oxo-2,3-dihydro-1*H*-indol-3yl)(2-phenylhydrazino)acetate (VIIIb). Yield 2.50 g (64%), mp 166–167°C. IR spectrum, v, cm<sup>-1</sup>: 3275 (C<sub>6</sub>H<sub>5</sub><u>NH</u>NH), 3140–3192 (<u>N<sup>1</sup>H</u>CO, C<sub>6</sub>H<sub>5</sub>NH<u>NH</u>), 1722 (C=O<sub>ester</sub>), 1688 (N<sup>1</sup>H<u>CO</u>), 1623, 1605, 1470. <sup>1</sup>H NMR spectrum, δ, ppm: 3.72 s (3H, COOC<u>H</u><sub>3</sub>), 3.86 s (3H, COOC<u>H</u><sub>3</sub>), 4.23 d, 4.38 d (2H, C<sup>3</sup>'H, C<sup>2</sup>H, *J* 4.6 Hz), 4.54 d, 4.73 d (2H, C<sup>3</sup>'H, C<sup>2</sup>H, *J* 5.8 Hz), 5.48 br.s (2H, 2C<sub>6</sub>H<sub>5</sub><u>NH</u>NH), 6.55 s (1H, C<sub>6</sub>H<sub>5</sub>NH<u>NH</u>), 6.75–7.48 group of signals (16H, 2C<sub>6</sub>H<sub>3</sub>, 2C<sub>6</sub>H<sub>5</sub>), 7.52 s (1H, C<sub>6</sub>H<sub>5</sub>NH<u>NH</u>), 10.15 s (1H, N<sup>1</sup>H), 10.47 s (1H, N<sup>1</sup>H). Found, %: C 52.12; H 3.97; Br 20.23; N 10.59. C<sub>17</sub>H<sub>16</sub>BrN<sub>3</sub>O<sub>3</sub>. Calculated, %: C 52.32; H 4.13; Br 20.48; N 10.77.

Methyl (1-acetyl-2-oxo-2,3-dihydroO-1*H*-indol-3yl)(2-phenylhydrazino)acetate (VIIIc). Yield 2.60 g (73%), mp 137–138°C. IR spectrum, v, cm<sup>-1</sup>: 3265 (C<sub>6</sub>H<sub>5</sub><u>NH</u>NH), 3135–3172 (<u>N</u><sup>1</sup>CO, C<sub>6</sub>H<sub>5</sub>NH<u>NH</u>), 1720 (C=O<sub>ester</sub>), 1685, 1673 (N<sup>1</sup><u>CO</u>), 1610, 1590, 1455. <sup>1</sup>H NMR spectrum, δ, ppm: 2.62 s (3H, C<u>H</u><sub>3</sub>CO), 2.66 s (3H, C<u>H</u><sub>3</sub>CO), 3.60 s (3H, COOC<u>H</u><sub>3</sub>), 3.73 s (3H, COOC<u>H</u><sub>3</sub>), 4.12 d, 4.24 d (2H, C<sup>3</sup>H, C<sup>2</sup>H, J 4.0 Hz), 4.35 d, 4.56 d 6.80 s (1H, C<sub>6</sub>H<sub>5</sub>NHNH), 6.84–7.48 group of signals

 $(18H, 2C_6H_4, 2C_6H_5), 7.58 \text{ s} (1H, C_6H_5NHNH)$ . Found, %: C 64.70; H 5.58; N 11.77. C<sub>19</sub>H<sub>19</sub>N<sub>3</sub>O<sub>4</sub>. Calculated, %: C 64.58; H 5.42; N 11.89.

(2H, C<sup>3</sup>H, C<sup>2</sup>H, J 5.2 Hz), 5.40 br.s (2H, 2C<sub>6</sub>H<sub>5</sub><u>NH</u>NH),

Ethyl (2-oxo-2,3-dihydro-1*H*-indol-3-yl)(2-phenylhydrazino)acetate (VIIId). Yield 2.63 g (81%), mp 150-151°C. IR spectrum, v, cm<sup>-1</sup>: 3293 (C<sub>6</sub>H<sub>5</sub>NHNH), 3135– 3182 (<u>N<sup>1</sup>H</u>CO, C<sub>6</sub>H<sub>5</sub>NH<u>NH</u>), 1732 (C=O<sub>ester</sub>), 1692 (N<sup>1</sup>H<u>CO</u>), 1615, 1604, 1470. <sup>1</sup>H NMR spectrum, δ, ppm: 1.20 t (3H, COOCH<sub>2</sub>C<u>H<sub>3</sub></u>), 1.26 t (3H, COOCH<sub>2</sub>C<u>H<sub>3</sub></u>), 3.82 q (2H, COOCH<sub>2</sub>CH<sub>3</sub>), 3.88 q (2H, COOCH<sub>2</sub>CH<sub>3</sub>), 4.16 d, 4.27 d (2H, C<sup>3</sup>H, C<sup>2</sup>H, J 4.5 Hz), 4.50 d, 4.66 d (2H, C<sup>3</sup>H, C<sup>2</sup>H, J 5.7 Hz), 5.44 br.s (2H, 2C<sub>6</sub>H<sub>5</sub>NHNH), 6.57 s (1H, C<sub>6</sub>H<sub>5</sub>NH<u>NH</u>), 6.75–7.32 group of signals (18H, 2C<sub>6</sub>H<sub>4</sub>, 2C<sub>6</sub>H<sub>5</sub>), 7.32 s (1H, C<sub>6</sub>H<sub>5</sub>NH<u>NH</u>), 10.47 s (1H, N<sup>1</sup>'H), 10.65 s (1H, N<sup>1</sup>'H). Found, %: C 66.52; H 5.67; N 13.11. C<sub>18</sub>H<sub>19</sub>N<sub>3</sub>O<sub>3</sub>. Calculated, %: C 66.45; H 5.89; N 12.91.

Ethyl (5-bromo-2-oxo-2,3-dihydro-1H-indol-3yl)(2-phenylhydrazino)acetate (VIIIe). Yield 2.63 g (68%), mp 158–159°C. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.18 t (3H, COOCH<sub>2</sub>C<u>H<sub>3</sub></u>), 1.23 t (3H, COOCH<sub>2</sub>C<u>H<sub>3</sub></u>), 3.77 q (2H, COOCH<sub>2</sub>CH<sub>3</sub>), 3.84 q (2H, COOCH<sub>2</sub>CH<sub>3</sub>), 4.11 d, 4.23 d (2H, C<sup>3</sup>'H, C<sup>2</sup>H, J 4.9 Hz), 4.46 d, 4.61 d (2H, C<sup>3</sup>H, C<sup>2</sup>H, J 6.2 Hz), 5.72 br.s (2H, 2C<sub>6</sub>H<sub>5</sub>NHNH), 6.43 s (1H, C<sub>6</sub>H<sub>5</sub>NH<u>NH</u>), 6.65–7.33 group of signals (16H, 2C<sub>6</sub>H<sub>3</sub>, 2C<sub>6</sub>H<sub>5</sub>), 7.69 s (1H, C<sub>6</sub>H<sub>5</sub>NH<u>NH</u>), 10.73 s (1H, N<sup>1</sup>H), 10.90 s (1H, N<sup>1</sup>H). Found, %: C 53.61; H 4.58; Br 19.87; N 10.52. C<sub>18</sub>H<sub>18</sub>BrN<sub>3</sub>O<sub>3</sub>. Calculated, %: C 53.48; H 4.49; Br 19.77; N 10.39.

Ethyl (1-acetyl-2-oxo-2,3-dihydro-1H-indol-3-yl)-(2-phenylhydrazino)acetate (VIIIf). Yield 2.60 g (70%), mp 144–145°C. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 2.55 s (3H, CH<sub>3</sub>CO), 2.63 s (3H, CH<sub>3</sub>CO), 1.22 t (3H, COOCH<sub>2</sub>CH<sub>3</sub>), 1.28 t (3H, COOCH<sub>2</sub>CH<sub>3</sub>), 3.78 q (2H, COOCH<sub>2</sub>CH<sub>3</sub>), 3.83 q (2H, COOCH<sub>2</sub>CH<sub>3</sub>), 4.15 d, 4.28 d (2H, C<sup>3</sup>'H, C<sup>2</sup>H, J 4.2 Hz), 4.38 d, 4.61 d (2H, C<sup>3</sup>'H, C<sup>2</sup>H, J 5.6 Hz), 5.53 br.s (2H, 2C<sub>6</sub>H<sub>5</sub>NHNH), 6.85 s (1H,  $C_6H_5NHNH$ , 6.90–7.57 group of signals (18H,  $2C_6H_4$ , 2C<sub>6</sub>H<sub>5</sub>), 7.71 s (1H, C<sub>6</sub>H<sub>5</sub>NH<u>NH</u>). Found, %: C 65.09; H 5.71; N 11.27. C<sub>20</sub>H<sub>21</sub>N<sub>3</sub>O<sub>4</sub>. Calculated, %: C 65.38; H 5.76; N 11.44.

## REFERENCES

- 1. Zhungietu, G.I., Oksindol i ego proizvodnye (Oxindole and Its Derivatives), Kishinev: Shtiintsa, 1973, p. 129.
- 2. Zhungietu, G.I., Dragalina, G.A., and Dorofeenko, G.N., Khim. Geterotsikl. Soedin., 1973, p. 40.
- 3. Otomasu, H., Tanaka, T., and Aoyagi, M., Chem. Pharm. Bull., 1976, vol. 24, p. 782.
- 4. Hassan, K.M. and Khalil, Z.H., J. Prakt. Chem., 1979, vol. 321, p. 870.
- 5. Joshi, K.C., Dandia, A., and Bhagat, S., Indian J. Chem., 1992, vol. 31A, p. 98; Ref. Zh. Khim., 1993, 12Zh398.
- 6. Al-Thebeiti, M.S., Heteroatom. Chem., 1994, vol. 5, p. 571; Chem. Abstr., 1995, vol. 123, 55734f.
- 7. Mogilaiah, K. and Rao, R.B., Indian J. Chem., 1998, vol. 37B, p. 139; Ref. Zh. Khim., 1999, 14Zh223.
- 8. Sharma, K., Jain, R., and Joshi, K.C., Indian J. Heterocycl. Chem., 1992, 1, 189; Chem. Abstr., 1992, vol. 117, 48259c.
- 9.Koz'minykh, V.O., Lomidze, K.Sh., Goncharov, V.I., Aksenov, A.V., Koz'minykh, E.N., and Berezin, A.N., Khim. Geterotsikl. Soedin., 2005, p. 792.
- 10. Koz'minykh, E.N., Aksenov, A.V., Koz'minykh, V.O., Lomidze, K.Sh., and Berezin, A.N., Azotsoderzhashchie geterotsikly (Heterocycles Containing Nitrogen), Kartsev, V.G., Ed., Moscow: ICSPF PRESS, 2006, vol. 2, p. 139.
- 11. Berezin, A.N., Koz'minykh, E.N., and Koz'minykh, V.O., Abstracts of Papers, Molodezhnaya nauchnaya shkolakonf. po organicheskoi khimii (Young Scientist Conference on Organic Chemistry), Ekaterinburg, 2002, p. 89.
- 12. Koz'minykh, V.O., Goncharov, V.I., Aksenov, A.V., Koz'minykh, E.N., Lomidze, K.Sh., and Berezin, A.N., Zh. Org. Khim., 2006, vol. 42, p. 1373.
- 13. Lomidze, K.Sh., Koz'minykh, E.N., Berezin, A.N., and Koz'minykh, V.O., Khimiya sinteticheskikh indol'nykh system (Chemistry of Synthetic Indol Systems), Kartsev, V.G., Moscow: IBS PRESS, 2004, vol. 3, p. 547.
- 14. Koz'minykh, V.O., Goncharov, V.I., Koz'minykh, E.N., and Lomidze, K.Sh., Khim. Geterotsikl. Soedin., 2006, p. 133.
- 15. Roth, H.J. and Lausen, H.H., Arch. Pharm., 1973, vol. 306, p. 767; Chem. Abstr., 1974, vol. 80, 27049 m.
- 16. Koz'minykh, E.N., Berezina, E.S., and Koz'minykh, V.O., Zh. Obshch. Khim., 1996, vol. 66, p. 1128.
- 17. Koz'minykh, E.N., Berezina, E.S., Kolla, V.E., Shelenkova, S.A., Voronina, E.V., and Koz'minykh, V.O., Khim.-Farm. Zh., 1997, vol. 31, no. 2, p. 31.
- 18. Osman, F.H. and El-Samahy, F.A., Phosph., Sulfur, Silicon. Relat. Elem., 1998, pp. 134-135, 437; Ref. Zh. Khim., 1999, 11Zh274.